

wherein m and n are coefficients equal to oxidation numbers of the anion A and B, respectively,

the anion A is selected from the group consisting of  $\text{Cl}^-$ ,  $\text{Br}^-$ ,  $\text{I}^-$ ,  $\text{F}^-$ ,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{PO}_4^{3-}$ ,  $\text{OH}^-$ ,  $\text{RCOO}^-$ , tartrate<sup>2-</sup>, citrate<sup>3-</sup> and an amino acid residue;

wherein R is selected from the group consisting of hydrogen, a  $\text{C}_1$ - $\text{C}_{20}$  straight chain hydrocarbon, a  $\text{C}_1$ - $\text{C}_{20}$  branched hydrocarbon and an aromatic group,

the colloidal cupric compound made by a process comprising the steps of:

purifying a  $\text{Cu}^{2+}$  solution by adding an oxidizing agent and  $\text{H}_3\text{PO}_4$  to the solution, and raising the pH of the solution.

9. (Twice Amended) A process for producing a colloidal cupric compound of formula (I):



wherein A and B are anions,

$$0 \leq x \leq 2,$$

$$0 < y \leq 2, \text{ and}$$

$$mx + ny = 2;$$

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